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Manual. If BC in Fig. 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , i.e., by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v A 0 2 \Delta = \Delta \Delta 0$
Eq. 1st.4 The slope of the vapor pressure curve of the solid solvent is also given by another form of the

LABORATORY MANUAL FOR PHYSICAL CHEMISTRY

Laboratory manual for physical chemistry where T_0 is the freezing point of the solvent If BC in Fig 1st-1 is taken to be a straight line, the slope dp_A/dT may be replaced by CB/BD , ie, by $(p_A - p_s)/\Delta T$, so that $p_p T H_p R T A s m v A 0 2 \Delta = \Delta \Delta 0$
Eq 1st4 The slope

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the temperature reaches a steady reading.
Record the temperature and the
equilibrium pressure. Increase the pressure
in the system by 3-5 mmHg (around 6
kPa) by allowing a small amount of air to
enter through tap B slowly.

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Laboratory Experiments and Laboratory Reports The course is built around several experiments in various topics in Physical Chemistry. Laboratory reports will be required for all experiments. Reports should be submitted on Thursday, the day you finish one experiment.

M.Sc. Physical Chemistry Laboratory
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Appropriate for junior-level courses, the modernized new edition of Halpern and McBane's physical chemistry lab manual offers an exceptional variety of tested and proven experiments—as well as detailed theoretical background material and clearly stated procedural objectives to help students understand the context and purpose of each procedure.

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Physical Chemistry Laboratory (CHM
423/423A)

Chem 343 Physical Chemistry Laboratory
Manual Spring 2014 – ADH 1/10/14 By
Audrey Dell Hammerich, Luke Hanley,
and Preston Snee with contributions from
Yoshitaka Ishii, Sanja Tepavcevic, and
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