

Fun With Predicting Reaction Products Worksheet Answers

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Fun With Predicting Reaction Products - Answers Predict the products of each of the following chemical reactions. If a reaction will not occur, explain why not: Before moving on, here are some general rules of thumb for how to figure out what will be made (and if the reaction will occur at all):

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probably a combustion reaction. The products will be CO₂ and H₂O. 2) If two elements or very simple molecules combine with each other, it's probably a synthesis reaction. The products will probably be predictable using the octet rule to find charges. 3) If one compound has an arrow coming off of it, it's probably a decomposition reaction. The products will either be a couple of very

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1) If something that has carbon and hydrogen reacts with oxygen, it's probably a combustion reaction. The products will be CO₂ and H₂O. 2) If two elements or very simple molecules combine with each other, it's probably a synthesis reaction. The products will probably be predictable using the octet rule to find charges.

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This Fun with Predicting Reaction Products worksheet also includes: Answer Key. Join to access all included materials. Predict the product of these chemical reactions! Using hints on one page, chemists analyze 8 sample chemical reactions, predicting the products of the given reactants. If a reaction will not occur, students explain why.

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The reactants suggest that this is a combustion reaction, meaning that the products must be carbon dioxide and water. Once you figure this out, the only thing left to do is balance it, as shown. Title: Fun With Predicting Reaction Products Author: Ian Guch Created Date:

Chapter 8 Practice Quiz Name per Show all work!!! Part 1 ...

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Fun With Predicting Reaction Products. Predict the products of each of the following chemical reactions. If a reaction will not occur, explain why not: 1) $\text{Ag}_2\text{SO}_4 + \text{NaNO}_3 \rightarrow \text{AgNO}_3 + \text{Na}_2\text{SO}_4$. 2) $\text{NaI} + \text{CaSO}_4 \rightarrow \text{CaI}_2 + \text{Na}_2\text{SO}_4$. 3) $\text{HNO}_3 + \text{Ca(OH)}_2 \rightarrow \text{H}_2\text{O} + \text{Ca(NO}_3)_2$. 4) $\text{Zn} + \text{HNO}_3 \rightarrow \text{Zn(NO}_3)_2 + \text{H}_2$

NOTES: CHEMICAL REACTIONS

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PREDICTING REACTION PRODUCTS: COMBUSTION REACTIONS! A hydrocarbon and oxygen (O_2) indicate a combustion reaction ! If it is combustion, then just write H_2O and CO_2 as products ! Then, balance the equation (can be tricky) !
Examples: $\text{CH}_4 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_4\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_2\text{H}_6 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$ $\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$

Steps to Predicting the Products of Chemical Reactions

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Predicting Products Of Chemical - Free Photos

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Predicting Reaction Products - Balance the equations and ... CHM 130 Predicting Products Worksheet Circle the appropriate reaction type for each, complete the reaction with products (remember to check charges for ionic compounds), include states, and finally balance each reaction. CB = combustion, AB = acid base neutralization, SR = single ...

Predicting Products Answer Key

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