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Gas Laws Part II Gay-Lussac's /u0026 Combined Gas Law ~~Combined Gas Law Worksheet Chart~~

Combined Gas Law Worksheet - Solutions. 1) If I initially have 4.0 L of a gas at a pressure of 1.1 atm, what will the volume be if I increase the pressure to 3.4 atm? $(1.1 \text{ atm})(4.0 \text{ L}) = (3.4 \text{ atm})(x \text{ L})$ $x = 1.29 \text{ L}$. 2) A toy balloon has an internal pressure of 1.05 atm and a volume of 5.0 L.

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Answers: COMBINED GAS LAW Remember to convert all temperatures to Kelvin. $P_1 V_1 T_1 = P_2 V_2 T_2$ 1.5 atm 3.0 L 20. C 293K 2.5 atm 1.9 L 30. C 303K 2 720 torr 256 mL 25 C 298 K 8.0x10² torr 250 mL 50. C 323 K 3 600. mmHg 2.5 L 22 C 295 K 760 mmHg 1.8 L 270 K 4 1.2 atm 750 mL 0.0 C 273.0 K 2.0 atm 500. mL 25 C 298 K 5 95 kPa 4.0 L

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3. A 3.25 L container of ammonia gas exerts a pressure of 652 mm Hg at a temperature of 243 K. Calculate the pressure of this same amount of gas in a 2.50 L container at a temperature of 221 K. 4. A sample of gas has a volume of 5.23 cm³ at a pressure of 72.6 kPa and a temperature of 25 ° C. What will be the volume of the gas if the pressure is

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October 3, 2019 June 7, 2019. Some of the worksheets below are Combined Gas Law Problems Worksheet Answer Key, Gas Laws Worksheet : Boyle ' s Law Problems, Charles ' Law Problems, Guy-Lussac ' s Law, Avogadros Law and Molar Volume at STP , Combined Gas Law Problems, Once you find your document (s), you can either click on the pop-out icon or download button to print or download your desired document (s).

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“ Gas Law ” Various Gas Laws Boyles Law: initial pressure equals final pressure times final volume $P_1 V_1 = P_2 V_2$ Charles Law: the ratio of volume to temperature of a given gas at fixed pressure is constant $V_1/T_1 = V_2/T_2$ Gay-Lussac ' s Law: the ratio of pressure to temperature of a given gas at fixed volume is constant $P_1/T_1 = P_2/T_2$ Avogadro's Law: at fixed pressure and temperature, the ratio of ...

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