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MASS and VOLUME. ATOMS and
MOLECULES the basic building blocks
of matter. ATOM-- the smallest unit of an

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element that still maintains the properties
that element. ELEMENT is a pure
substance made up of only one type of
atom.

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25 Electrons in Atoms Reviewing

Vocabulary Match the definition in

Column A with the term in Column B.

Column A Column B 1. The set of

frequencies of the electromagnetic waves

emitted by the atoms of an element 2. The

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minimum amount of energy that can be
lost or gained by ...

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Stoichiometry (that is, energy produced
or absorbed) A system that can exchange
neither energy nor matter with its

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extensive property. intensive property.
extensive properties (examples) intensive

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property (examples) a property that depends on the amount of matter in a sample. a property that depends on the type of matter in a sample not. mass, volume, length, weight.

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11 31 number of moles of known

substance C H C H C H O O O CO CO

CO H O H O H O number of moles of

unknown substance O CO H O C H CO H

O C H O H O C H O CO use mole ratio 5

mol O 2 1 mol C 3 H 8 3 mol CO 2 1 mol

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C_3H_8 4 mol H_2O 1 mol C_3H_8 1 mol
 C_3H_8 5 mol O_2

MATH SKILLS TRANSPARENCY MASTER 16

A **chemical change** is a change that produces matter with a different composition than the original matter.

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Chemical Change A change in which one or more substances are converted into different substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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