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~~a) O atom = 0, C atom = 0, and N atom = 0 b) O atom = 0, C atom = 0, and N atom = - c) O atom = -1, C atom = 0, and N atom = 0 d) O atom = -1, C atom = -1, and N atom = + e) O atom = +1, C atom = 0, and N atom = - . 17. Use VSEPR to predict the electron-pair geometry and the molecular geometry of sulfur dioxide, SO₂.~~

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~~that portion of chemistry dealing with numerical relationships in chemical reactions; the calculation of quantities of substances involved in chemical equations~~

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