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Acids and Bases in Water  
Water (H<sub>2</sub>O) - the most  
important molecule on  
earth. Even in pure water,  
there are small amounts of

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ions from the equilibrium below ("self-ionization of water" or "auto-ionization).

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Answers Acids And Bases

model of acids and bases states that an acid contains

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the element hydrogen and forms ions of this element when it is dissolved in water. a base contains the hydroxide group and dissociates to produce hydroxide ions in aqueous solution



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typical reaction, if the reacting acid is strong, the base produced is strong.

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Double-replacement rxn Rxn  
in which an acid and a base  
in an aqueous solution react  
to produce a

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Bases Section 118.18.1

Figure 18.1 Rhododendrons flourish in rich, moist soil that is moderately acidic, whereas semper-vivum, commonly called hen and chicks, grows best in drier, slightly basic soil.

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Acids are named by changing the ending of the alkane to -oic acid. i.e. a 4 carbon acid goes from butane to butanoic acid Acid salts are named by changing the acid ending to -oate i.e the 4 carbon acid goes from

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butanoic acid to sodium butanoate Esters are named by changing the ending of the alkane to -oate and by using alkyl name i.e. the 4 carbon ester below is methyl butanoate (methyl for  $\text{OCH}_3$  portion)  $\text{O} \text{O}$  Anhydrides are



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dehydrated esters that  
replace -oic acid with word  
anhydride Acyl ...

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Reactions. acid,  $H_3PO_4$ ,  
are triprotic acids. Most of  
the phosphoric acid produced  
by the chemical industry is  
used to make fertilizers and  
detergents, but it is also

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used to make  
pharmaceuticals,

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