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Stoichiometry 1.  $1 \text{ ZnI}_2 \rightarrow 1 \text{ Zn} + 1 \text{ I}_2$  How many grams of iodine will you produce if you begin your rxn with 56.7g of zinc iodide? 56.7 g 1 mole 1 mole 253.8 g 319.19 g 1 mole 1 mole = 45.08 grams of Iodine 2. 1



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between the amounts of reactants used and amounts of products formed by a chemical reaction is called stoichiometry.

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19 - 12/3 Finish Lab – Iron and

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