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Why do atoms form molecules? The quantum physics of chemical bonds explained
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Atoms and Bonding Chapter Test A Multiple Choice Write the letter of the correct
answer on the line at the left. ____ 1. Which is a property shared by most molecular
compounds? a. high boiling point b. high melting point c. low melting point d. nonpolar
bonds ____ 2. When an atom loses an electron, it becomes a a. positive ion. b.
negative ion.

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~~Atoms and Bonding~~

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Chapter 5 Atoms and Bonding Chapter Preview Questions 4. Compounds are formed by a. combining two or more different elements. b. bombarding atoms with high-speed particles. c. combining two or more different nuclei. d. dissolving a solid in a

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liquid.

~~Chapter 5 Atoms and Bonding~~

What kind of bonds do these compounds have? a. ionic bonds b. covalent bonds c. metallic bonds ____ 24. Applying Concepts Which compound has a polar bond? a. HF b. F₂ c. O₂ ____ 25. Applying Concepts Which compound has a triple bond? a. O₂ b. F₂ c. N₂ H (+) (-) F O O N N F F Atoms and Bonding

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chemistry chapter 9 covalent bonding test answers, Covalent Bonding: Orbitals 9 One of the most useful parts of this model is its ability to accurately predict para- and diamagnetism as well as bond order. This device is used to test the paramagnetism of homonuclear samples. When the electromagnet is on, a paramagnetic substance is drawn down into it and appears heavier on the balance.

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Atoms & Bonding - Chapter Summary. This chapter outlines atoms and bonding as they're described in chemistry, including how atoms are defined and the ways that compounds and molecules are made up ...

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