

Atomic Structure Worksheet With Answers

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Atomic Structure Worksheet - Solution Key. Fill in the blanks for the elements in this chart. For the purposes of this chart, round all atomic masses to the nearest whole number. Element Number of Protons Number of Neutrons Number of Electrons Atomic Mass Atomic Number. lithium 3 4 3 7 3.

Atomic Structure Worksheet - Solution Key

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Atomic Structure Worksheet worksheet

The atomic number tells you the number of in one atom of an element. It also tells you the number of in a neutral atom of that element. The atomic number gives the "identity" of an element as well as it's location on the Periodic Table. No two different elements will have the atomic number.

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Atomic Structure Worksheet. The atomic number of an element is the number of protons in the nucleus of an atom. The mass number is the total number of protons plus neutrons in the nucleus of an atom. Use the periodic table to help you complete the chart below. Remember to round the atomic mass to the nearest whole number to get the mass number if the number of neutrons is not given.

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1. In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

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ATOMIC STRUCTUREName An atom is made up of protons and neutrons (both found in the nucleus) and electrons (in the surrounding electron cloud). The atomic number is equal to the number of protons, The mass number is equal to the number of protons plus neutrons. In a neutral atom, the number of protons equals the number of electrons.

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