

## 9 2 Practice Problems Chemistry Answer Key

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Practice Problems: 1. If 8 out of 40 patients suffered serious side effects from a new medication, is there sufficient evidence at the 1% level of significance to claim that the true proportion is not equal to 0.40? Hypotheses: HO: p = 0.40 HA: p  $\neq$  0.40 Test Statistic:  $\hat{p}$  - 0.20 0.40 2.5820 0.40 0.60 40 z -

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Name Date (D O 13—2 Practice Problems 1. 2. 3. 4. 5. 6. 8. 9. The air pressure for a certain tire is 109 kPa. What is this pressure in atmospheres?

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Liquid A. (0.75 grams per milliliter) 8. 68 kilograms 9. 8.33 pounds (2.2 kilograms = 1 pound, 1 liter = 0.264 gallons) 10. 483.6 centimeters<sup>3</sup> Tips for Answering Density Questions When you're asked to calculate density, make sure your final answer is given in units of mass (such as grams, ounces, pounds, kilograms) per volume (cubic ...

Density Test Questions with Answers

Practice Problems: Stoichiometry (Answer Key) Balance the following chemical reactions: a. 2 CO + O<sub>2</sub> → 2 CO<sub>2</sub> b. 2 KNO<sub>3</sub> → 2 KNO<sub>2</sub> + O<sub>2</sub> c. 2 O<sub>3</sub> → 3 O<sub>2</sub> d. NH<sub>4</sub>NO<sub>3</sub> → N<sub>2</sub>O + 2 H<sub>2</sub>O e. 4 CH<sub>3</sub>NH<sub>2</sub> + 9 O<sub>2</sub> → 4 CO<sub>2</sub> + 10 H<sub>2</sub>O + 2 N<sub>2</sub> f. Cr(OH)<sub>3</sub> + 3 HClO<sub>4</sub> → Cr(ClO<sub>4</sub>)<sub>3</sub> + 3 H<sub>2</sub>O Write the balanced chemical equations of each reaction:

Practice Problems: Stoichiometry - Department of Chemistry

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Chemistry Packet Answers 9 2 Practice Problems

Balancing Equations: Answers to Practice Problems 1. Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers).

Balancing Equations: Practice Problems

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Practice Chemistry with Worked Chemistry Problems

13.1A: Keto-enol tautomerization. When we draw a ketone or aldehyde using the Lewis structure convention, we show a double bond between the carbonyl carbon and the oxygen - this is known as the keto form. It turns out that ketones and aldehydes often exist in rapid equilibrium with a tautomeric form known as an enol. The alpha-protons of carbonyls, as you recall, are somewhat acidic: the pK<sub>a</sub> ...

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